

HYDROGENOLYSIS OF THE DIOXOLAN TYPE EXO- AND ENDO- BENZYLIDENE
DERIVATIVES OF CARBOHYDRATES WITH THE $\text{LiAlH}_4\text{-AlCl}_3$ REAGENT

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The active AlH_2Cl reagent formed from equimolar quantities of AlCl_3 and LiAlH_4 , exhibits the properties of both Lewis acids and hydride donors¹. This is the basis of its reactions with acetals to yield hydroxyethers^{2,3,4}.

The direction of the ring cleavage of the 4,6-O-benzylidene acetals of hexosides is determined by the steric requirements of the substituents near the dioxan skeleton^{5,6,7}.

The secondary *cis* axial-equatorial hydroxyl groups of pyranosides react with benzaldehyde to give 2-phenyl-1,3-dioxolan derivatives existing in two relatively easily separable diastereomeric forms, absolute configurations of which can be determined by n.m.r.^{8,9}.

The following four pairs of isomers were synthesized and subjected to hydrogenolysis (in ether- CH_2Cl_2 /1:1/, 1.1 mols of reagent, at 45°). All percentage values show product ratios and not the isolated yields.

Benzyl exo-3,4-O-benzylidene- β -D-arabinopyranoside (1, mp 144-145°; $[\alpha]_D = -148^\circ$; $\delta_{\text{CH-Ph}} = 6.18$)¹⁰ gave 70 % benzyl 3-O-benzyl- β -D-arabinopyranoside isolated as the diacetate (9, syrup, $[\alpha]_D = -159^\circ$) and 30 % benzyl 4-O-benzyl- β -D-arabinopyranoside, isolated as its crystalline diacetate (10, mp 66-68°; $[\alpha]_D = -198^\circ$). Benzyl endo-3,4-O-benzylidene- β -D-arabinopyranoside (2, mp 96-97°; $[\alpha]_D = -171^\circ$; $\delta_{\text{CH-Ph}} = 5.78$) yielded, upon hydrogenolysis and acetylation 9 and 10 in a ratio of 22:78. Saponification of 9 and 10 gave a periodate-resistant and a periodate-oxidizable product, respectively.

Hydrogenolysis of benzyl 2-O-benzyl-exo-3,4-O-benzylidene- β -D-arabinopyranoside (3, mp 85-86°; $[\alpha]_D = -103^\circ$; $\delta_{\text{CH-Ph}} = 6.02$) gave 81 % benzyl 2,3-di-O-benzyl- β -D-arabinopyranoside (11, syrup, $[\alpha]_D = -75^\circ$) and 19 % benzyl 2,4-

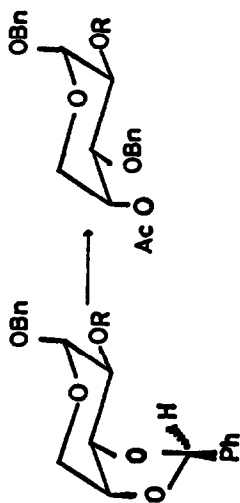
di-O-benzyl- β -D-arabinopyranoside (12, mp 73-74 $^{\circ}$; $[\alpha]_D = -123^{\circ}$). 11 and 12 were obtained from benzyl 2-O-benzyl-endo-3,4-O-benzylidene- β -D-arabinopyranoside (4, mp 80-82 $^{\circ}$; $[\alpha]_D = -164^{\circ}$; $\delta_{\text{CH-Ph}}^{\text{H}}$: 5.89) in a ratio of 16:84.

The ring cleavage of phenyl 2-O-benzyl-exo-3,4-O-benzylidene- β -D-fucopyranoside (5, mp 93-94 $^{\circ}$; $[\alpha]_D = +13^{\circ}$; $\delta_{\text{CH-Ph}}^{\text{H}}$: 6.00) yielded 71 % phenyl 2,3-di-O-benzyl- β -D-fucopyranoside (13, mp 89-90 $^{\circ}$; $[\alpha]_D = +12^{\circ}$) and 29 % phenyl 2,4-di-O-benzyl- β -D-fucopyranoside (14, mp 51-53 $^{\circ}$; $[\alpha]_D = +9^{\circ}$). Phenyl 2-O-benzyl-endo-3,4-O-benzylidene- β -D-fucopyranoside (6, mp 103-106 $^{\circ}$; $[\alpha]_D = +46^{\circ}$; $\delta_{\text{CH-Ph}}^{\text{H}}$: 5.90) gave, in quantitative yield, 14, only. The only product from benzyl 2,6-di-O-benzyl-exo-3,4-O-benzylidene- β -D-galactopyranoside (7, mp 132-133 $^{\circ}$; $[\alpha]_D = -2.6^{\circ}$; $\delta_{\text{CH-Ph}}^{\text{H}}$: 5.95) was benzyl 2,3,6-tri-O-benzyl- β -D-galactopyranoside (15, syrup, $[\alpha]_D = -14^{\circ}$). The cleavage of the dioxolan ring of benzyl 2,6-di-O-benzyl-endo-3,4-O-benzylidene- β -D-galactopyranoside (8, mp 65-66 $^{\circ}$; $[\alpha]_D = +12^{\circ}$; $\delta_{\text{CH-Ph}}^{\text{H}}$: 5.85) gave benzyl 2,4,6-tri-O-benzyl- β -D-galactopyranoside (16, mp 60-63 $^{\circ}$; $[\alpha]_D = -16^{\circ}$). The reactions of both 7 and 8 are highly stereoselective, i.e. in 7 the only site of the attack by the reagent is the axial C₄-oxygen atom of the galactopyranoside which also participates in the the dioxolan ring. However, in the case of 8 the reagent attacks only the equatorial C₃-oxygen atom.

The structure of the products of the cleavage reactions was proved by the methyl ethers obtained upon successive methylation, catalytic hydrogenation and acid hydrolysis of them.

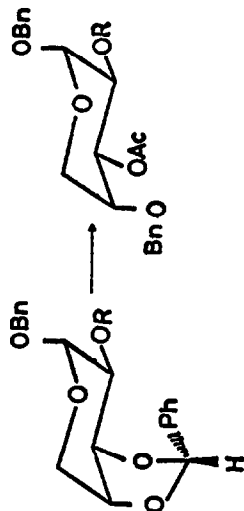
These reactions prove that the hydrogenolytic ring cleavages of 3,4-O-benzylidene pyranosides proceed via different routes. In the exo isomers the reagent attacks the axial oxygen atom forming an axial hydroxyl-equatorial benzyl derivative. On the contrary, in the endo isomers the equatorial oxygen atom is attacked leading to a product with the hydroxyl group in the equatorial and the O-benzyl group in the axial position, respectively.

Our results suggest that both 1 and 2 are ${}^1\text{C}_4$ conformers. 9 can be derived from 1, after hydrogenolysis and acetylation, if the C₄-O in 1 is axial and 10 from 2 if the C₃-O is equatorial. The n.m.r. spectra of both 1 and 2 show that they are strongly deformed ${}^1\text{C}_4$ conformers ($J_{4,5a} \approx J_{4,5e} \approx 1.8 \text{ Hz}$)



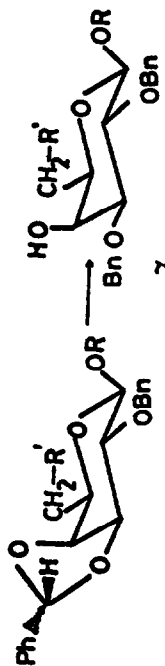
1 R = H
2 R = Bn

9 R = Ac
11 R = Bn



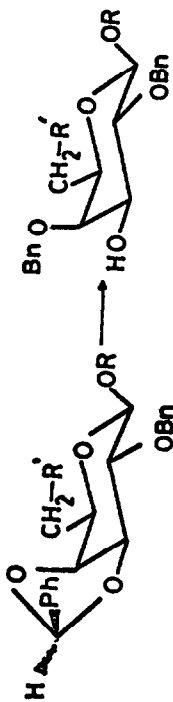
3 R = H
4 R = Bn

10 R = Ac
12 R = Bn



5 R = Ph; R' = H
7 R = Bn; R' = OBn

13 R = Ph; R' = H
15 R = Bn; R' = OBn



6 R = Ph; R' = H
8 R = Bn; R' = OBn

14 R = Ph; R' = H
16 R = Bn; R' = OBn

Bn = Benzyl
Ac = Acetyl

and this finding is confirmed by other authors⁹, too.

This procedure, offers a new method for the selective protection of cis axial-equatorial hydroxyl groups of hexo- and pentopyranosides, in form of benzyl ether.

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- 10 Optical rotations were measured in CHCl_3 . N.m.r. spectra were recorded on a Jeol 100 MHz instrument in CDCl_3 . The ratio of the products was determined on a Hewlett-Packard 5830 A gas chromatograph using a UCCW-982 (3 %) column.